

Geographic Variation of Mercury and Other Elements in U.S. Coal

Jeffrey Quick¹, D. Tabet¹, S. Wakefield¹,
R. Bon¹, T. Brill²; ¹ Utah Geological Survey,
and ² Utah Division of Public Utilities,
Salt Lake City, UT, USA

Air Quality V
18 - 21 September 2005
Marriott Crystal Gateway
Arlington, VA

U.S. Department of Energy
National Energy Technology Laboratory
Agreement DE-FG26-03NT41901
Sara M. Pletcher, Manager

Objective Use available coal quality data to illustrate which coals can be used with existing technology to reduce mercury emissions.

1

Data sources

ICR 2 (1999) 25,825 records

epa.gov/ttn/atw/combust/utitox/utoxpg.html

FERC 423 (1999) 19,507 records (also 2002)

eia.doe.gov/cneaf/electricity/page/ferc423.html

CTRDB (1992 - 1999) 5,823 records

eia.doe.gov/cneaf/coal/ctrdb/database.html

COALQUAL (1973 - 1989) 5,059 records

Bragg, L.J., and others 1997, U.S. Geological Survey Open File Report 97-134.

MSHA (1999) 1,342 records

msha.gov/STATS/PART50/P50Y2K/A&I/1999/caim1999.exe

EIA 423 (2002) 584 records

eia.doe.gov/cneaf/electricity/page/eia423.html

EIA 906-920 (2004) 476 records

eia.doe.gov/cneaf/electricity/page/eia906_920.html

ICR 3 (1999) 240 records

epa.gov/ttn/atw/combust/utitox/utoxpg.html

and analysis of ICR 3 data by:

SAIC (2003) <netl.doe.gov/coal/E&WR/mercury/pubs/DOE_Report_v120803.pdf>

ENSR (2003) <epa.gov/ttn/atw/combust/utitox/final_ensr_multivar.pdf>

Roberson (2002) <epa.gov/ttn/atw/combust/utitox/epavarifnl.doc>

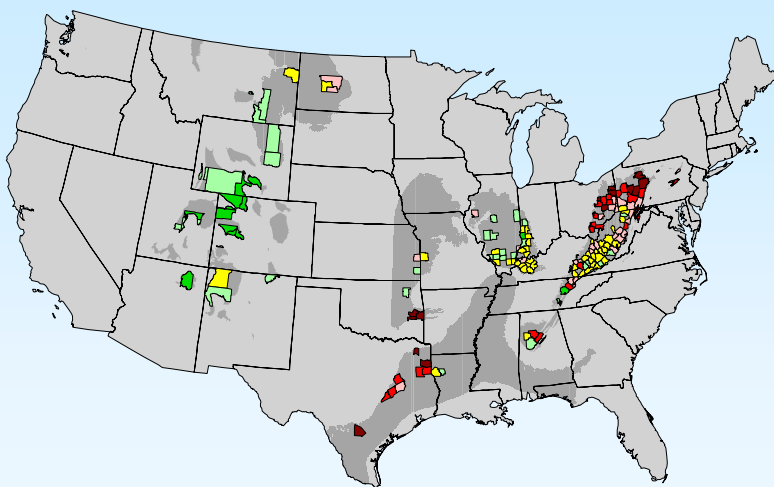
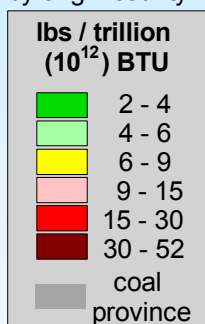
CRDB (1997) <eia.doe.gov/cneaf/coal/reserves/database.html>

2

Coal quality maps

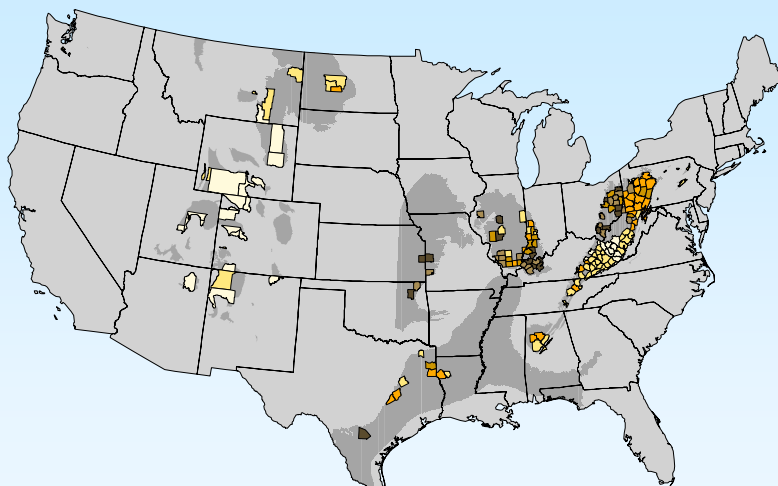
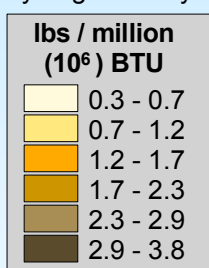
Mercury

ICR 2 data
commercial coal
by origin county



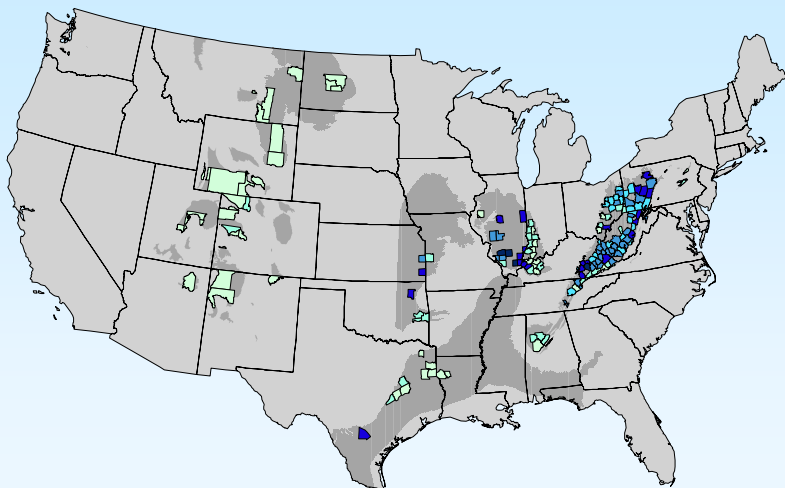
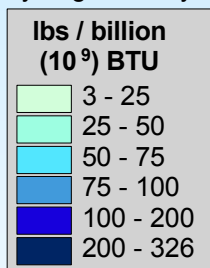
Sulfur

FERC 423 data
commercial coal
by origin county



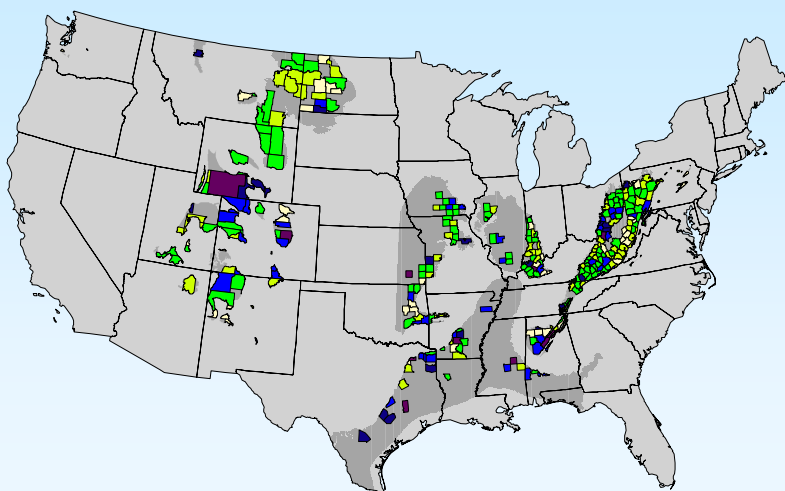
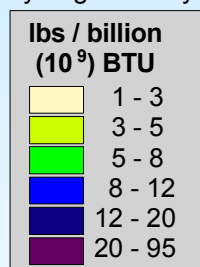
Chlorine

ICR 2 data
commercial coal
by origin county



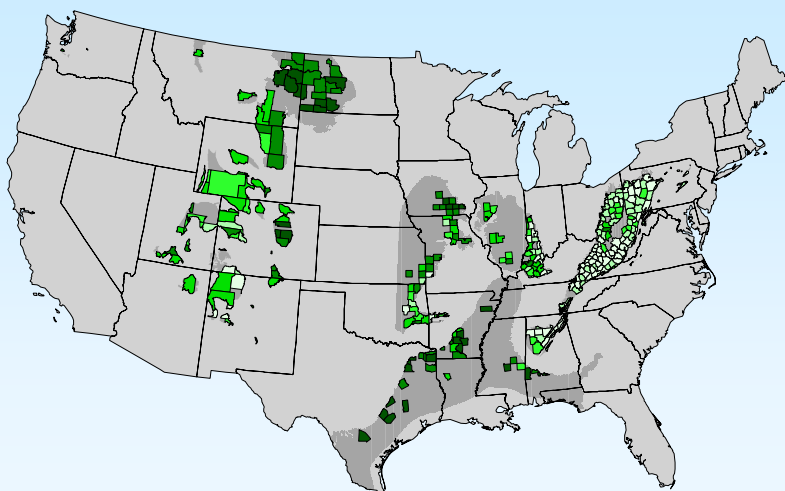
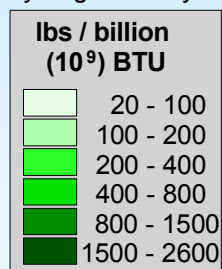
Fluorine

COALQUAL data
in-ground coal
by origin county



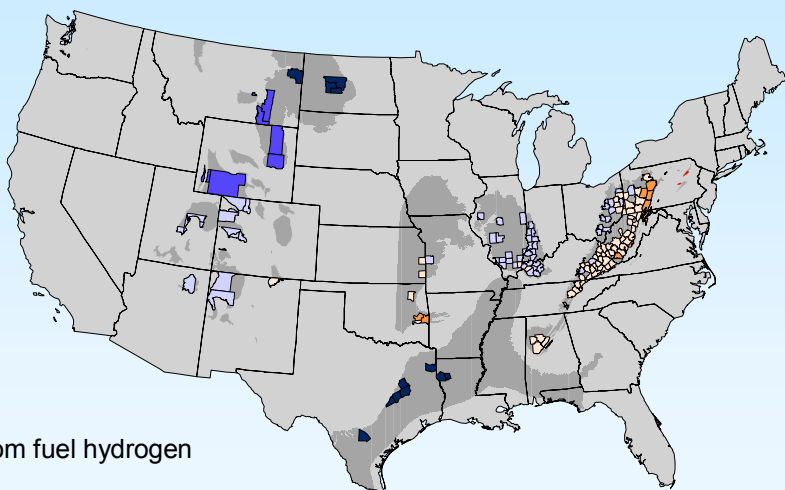
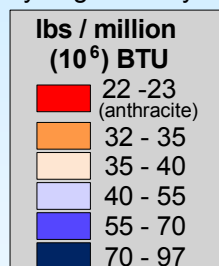
Calcium

COALQUAL data
in-ground coal
by origin county



Moisture*

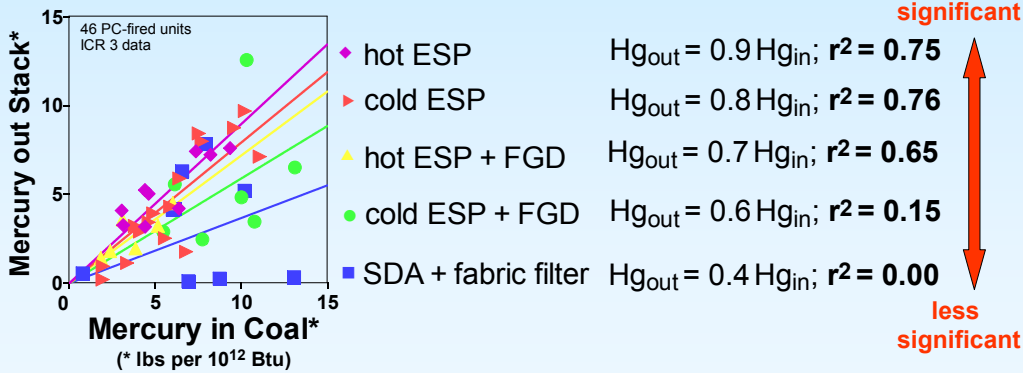
estimated,
commercial coal
by origin county



* includes H₂O from fuel hydrogen

3

The significance of the coal mercury content depends on the emission control technology.

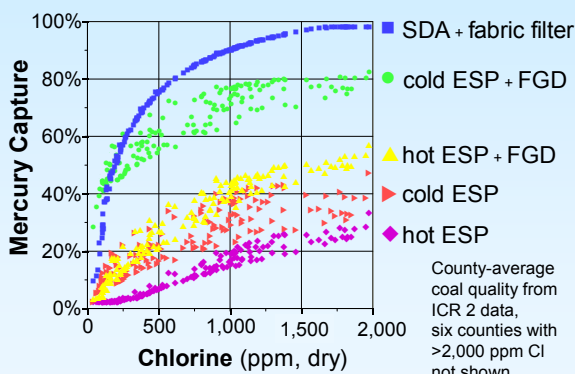


4

The average result of equations that predict mercury capture - applied to coal from 162 U.S. counties.

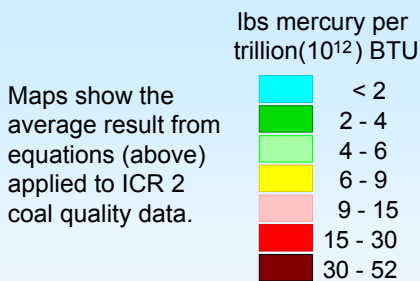
TECHNOLOGY	Equation to Predict Mercury Capture (100% capture = 1)	r ²	n
cESP			
Roberson (2002)	$0.1133 \ln \left(\frac{Cl_{avg}}{100 Cl_{min}} \right) - 0.2687$	0.53	28
model 2, SAIC (2003)	$1 - Exp \left(-7.33E^{-4} - 3.309 \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.47	12
model 3, SAIC (2003)	$1 - Exp \left(1.6374 - 0.18693 \ln \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.38	12
cESPFGD			
Roberson (2002)	$0.1137 \ln \left(\frac{Cl_{avg}}{100 Cl_{min}} \right) - 0.1428$	0.70	11
model 1, SAIC (2003)	$1 - Exp \left(0.8529 - 0.27149 \ln \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.74	8
model 3, SAIC (2003)	$1 - Exp \left(-0.2559 - 2.3343E^{-4} \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.73	8
hESP			
model 1, SAIC (2003)	$1 - Exp \left(0.9451 - 9.995E^{-4} \ln \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.42	7
model 2, SAIC (2003)	$1 - Exp \left(0.0611 - 2.165E^{-4} \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.54	7
model 3, SAIC (2003)	$1 - Exp \left(0.1234 - 1.024E^{-4} \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.39	9
hESPFGD			
model 1, SAIC (2003)	$1 - Exp \left(2.7919 - 0.29952 \ln \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.75	6
model 2, SAIC (2003)	$1 - Exp \left(-3.59E^{-4} - 9.595E^{-4} \ln \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.67	6
model 4, SAIC (2003)	$1 - Exp \left(2.9618 - 0.2684 \ln \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.42	6
SDAFF			
Roberson (2002)	$0.2864 \ln \left(\frac{Cl_{avg}}{100 Cl_{min}} \right) - 1.1393$	0.91	10
model 1, SAIC (2003)	$1 - Exp \left(10.7111 - 1.22628 \ln \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.89	10
ENSER (2003)	$1 - Exp \left(-0.19992 - 2.164E^{-4} \left(\frac{100 Cl_{avg}}{Cl_{min}} \right) \right)$	0.94	10

equations derived from ICR 3 data

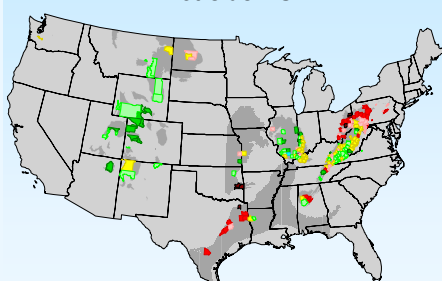


5

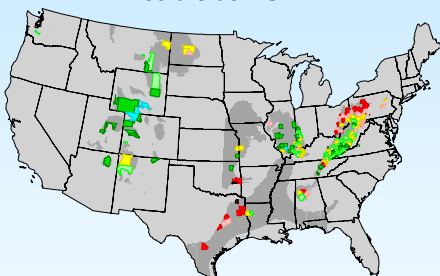
Mercury emissions for existing technology by coal origin



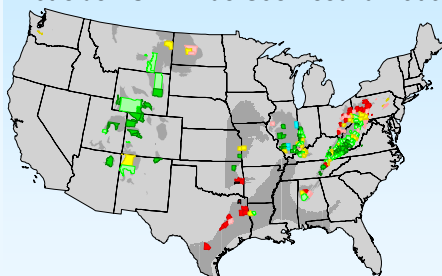
hot-side ESP



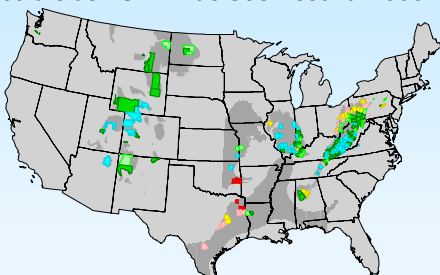
cold-side ESP



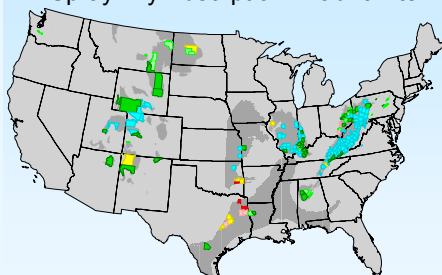
hot-side ESP + Flue Gas Desulfurization



cold-side ESP + Flue Gas Desulfurization



Spray Dry Adsorption + fabric filter



6

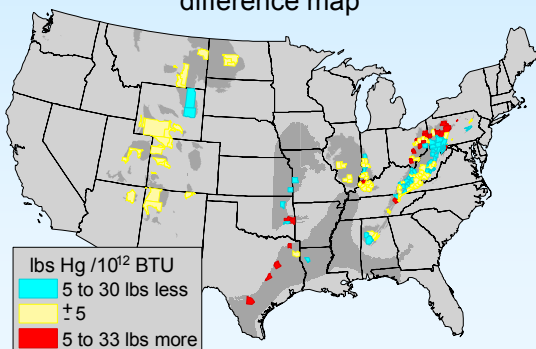
Commercial U.S. coal contains less mercury than the in-ground coal but not everywhere !

11 lbs Hg/10¹² BTU in-ground coal
(COALQUAL Hg data, CRDB tonnage)

8.3 lbs Hg/10¹² BTU commercial coal
(ICR 2 Hg data, MSHA tonnage)

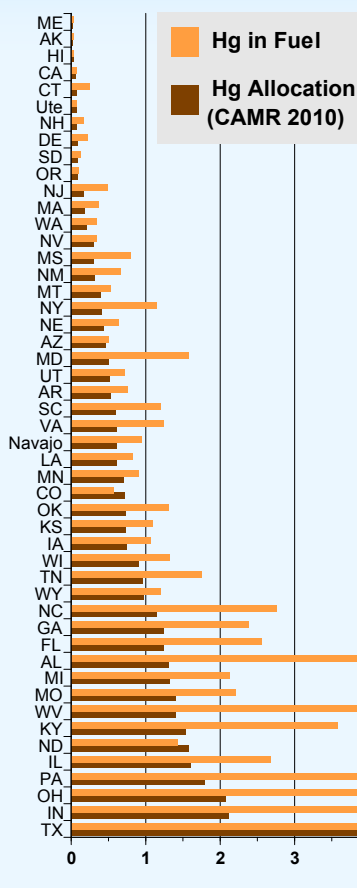
Red areas show where commercial coal has more mercury than the in-ground coal resource.

commercial minus in-ground coal Hg difference map

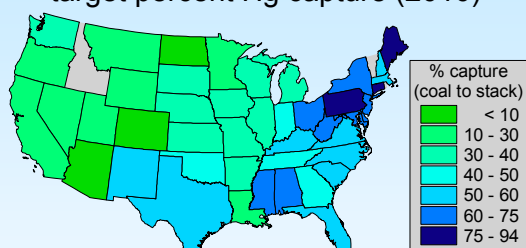


7

2010 CAMR emission targets

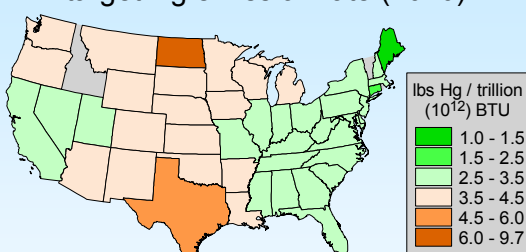


target percent Hg capture (2010)



No CAMR affected units in Idaho or Vermont (grey)

target Hg emission rate (2010)



No CAMR affected units in Idaho or Vermont (grey)

1999 county-average Hg values (ICR 2 data)
2002 state-average fuel origins (FERC and EIA 423 data)
2004 fuel consumption levels (mostly EIA 906/920 data)
2005 Clean Air Mercury Rule (CAMR)

Assumes no trading, no banking, and no change to the amount, origin, or quality of coal.

8

Ways to reduce mercury emissions

500 to 1,000 ppm fuel chlorine: cold-side ESP + FGD
SDA + fabric-filter

low-mercury coal: hot-side ESP
cold-side ESP

selective mining or washing: parts of Pennsylvania,
Ohio, and Texas

website: <http://geology.utah.gov/emp/mercury/index.htm>
comments, corrections, questions, ideas, data... welcome!

jeffreyquick@utah.gov

This poster was prepared with the support of the U.S. Department of Energy, under Award No. DE-FG26-03NT41901. However, any opinions, findings, conclusions, or recommendations expressed herein are those of the author and do not necessarily reflect the views of the DOE.

